

See discussions, stats, and author profiles for this publication at: https://www.researchgate.net/publication/269189363

The Discoloration of the Taj Mahal due to Particulate Carbon and Dust Deposition

Article in Environmental Science & Technology · December 2014

DOI: 10.1021/es504005q · Source: PubMed

CITATIONS	READS
11	230

9 authors, including:



Mike Bergin

Georgia Institute of Technology 36 PUBLICATIONS 809 CITATIONS

SEE PROFILE



Sachchida Nand Tripathi Indian Institute of Technology Kanpur 223 PUBLICATIONS 4,253 CITATIONS

SEE PROFILE



Tarun Gupta

Indian Institute of Technology Kanpur

139 PUBLICATIONS **1,327** CITATIONS

SEE PROFILE

Some of the authors of this publication are also working on these related projects:



All content following this page was uploaded by J. Jai Devi on 20 December 2014.

The user has requested enhancement of the downloaded file. All in-text references <u>underlined in blue</u> are added to the original document and are linked to publications on ResearchGate, letting you access and read them immediately.



Article

Subscriber access provided by City University of Hong Kong Library

The Discoloration of the Taj Mahal due to Particulate Carbon and Dust Deposition

Mike H. Bergin, Sachi Nand Tripathi, J Jai Devi, Tarun Gupta, Michael McKenzie, K S Rana, Martin M. Shafer, Ana M Villalobos, and James J. Schauer *Environ. Sci. Technol.*, Just Accepted Manuscript • DOI: 10.1021/es504005q • Publication Date (Web): 03 Dec 2014 Downloaded from http://pubs.acs.org on December 10, 2014

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



Environmental Science & Technology is published by the American Chemical Society. 1155 Sixteenth Street N.W., Washington, DC 20036

Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties.



The Discoloration of the Taj Mahal due to Particulate Carbon and Dust Deposition

M.H. Bergin^{1,2*}, <u>S.N. Tripathi</u>^{3*}, J. Jai Devi^{1,2}, T. Gupta³, M.Mckenzie², K.S. Rana⁴, M.M.
Shafer⁵, Ana M. Villalobos⁵, J.J. Schauer⁵

⁵ ¹School of Earth and Atmospheric Sciences, Georgia Institute of Technology, Atlanta, GA, USA.

⁶ ²School of Civil and Environmental Engineering, Georgia Institute of Technology, Atlanta, GA,

7 USA, mike.bergin@ce.gatech.edu, 404-894-9723

³Department of Civil Engineering and Center for Environmental Science and Engineering, Indian

9 Institute of Technology, Kanpur, India, snt@iitk.ic.in, 91-512-259-7845

⁴Archaelogical Survey of India (Science Branch), Delhi, India.

⁵Environmental Chemistry and Technology Program, University of Wisconsin at Madison,

12 Madison, WI, USA.

13 KEYWORDS: Brown Carbon (BrC), Black Carbon (BC), Dust, Surface Discoloration, Biomass

14 Burning, Trash/Refuse Burning

16 Abstract

17 The white marble domes of the Taj Mahal are iconic images of India that attract millions of 18 visitors every year. Over the past several decades the outer marble surfaces of the Taj Mahal 19 have begun to discolor with time and must be painstakingly cleaned every several years. 20 Although it has been generally believed that the discoloration is in some way linked with poor air 21 quality in the Agra region, the specific components of air pollution responsible have yet to be 22 identified. With this in mind, ambient particulate matter (PM) samples were collected over a one-23 year period and found to contain relatively high concentrations of light absorbing particles that 24 could potentially discolor the Taj Mahal marble surfaces, that include black carbon (BC), light 25 absorbing organic carbon (brown carbon, BrC), and dust. Analyses of particles deposited to 26 marble surrogate surfaces at the Taj Mahal indicate that a large fraction of the outer Taj Mahal 27 surfaces are covered with particles that contain both carbonaceous components and dust. We 28 have developed a novel approach that estimates the impact of these deposited particles on the 29 visible light surface reflectance, which is in turn used to estimate the perceived color by the 30 human eye. Results indicate that deposited light absorbing dust and carbonaceous particles (both 31 BC and BrC from the combustion of fossil fuels and biomass) are responsible for the surface 32 discoloration of the Taj Mahal. Overall, the results suggest that the deposition of light absorbing 33 particulate matter in regions of high aerosol loading are not only influencing cultural heritage but 34 also the aesthetics of both natural and urban surfaces.

36 Introduction

On the timescale of several years the outer marble surfaces of the Taj Mahal become discolored and must be cleaned in a time consuming process (*1*). Figure 1 shows a cleaned section of a marble Mosque Dome at the Taj Mahal (where cleaning involves applying a layer of clay and removing the clay after it

41 dries, followed by rinsing 42 the surface with clean 43 water) directly next to an 44 area being prepared for 45 cleaning. An obvious 46 contrast is seen between 47 the clean, white marble 48 surface and the darker 49 discolored surface of the 50 Marble Dome. Many 51 have been measures



FIGURE 1. A clean surface and an area being prepared for cleaning on a marble Mosque dome at the Taj Mahal

undertaken to avoid the impact of local air pollution, including restricting traffic within 1 km of the grounds and limiting the emissions of industrial pollution in the city of Agra, where the Taj Mahal is located. While detailed scientific studies have not been reported in the literature, past efforts focusing on the discoloration have hypothesized that local air quality is responsible and suggestions have included surface reactions with gas-phase SO₂, as well as aqueous phase chemistry linked with the deposition of fog droplets, and water condensation (*2*), as well as dust deposition (1). Despite efforts to keep the outer surfaces of the Taj Mahal white, it continues to
become discolored with time, and the reason for the discoloration is not currently understood.

60 Recent work has reported poor air quality throughout the Indo-Gangetic plain (3,4), including 61 relatively high concentrations of particulate matter in Agra (5-7). Particulate matter in the region 62 includes the light absorbing components black carbon (BC), light absorbing organic carbon (a 63 fraction of which can absorb light preferentially in the UV region and is often termed brown 64 carbon, BrC), and dust (8). Both organic carbon and dust have the potential to preferentially 65 absorb solar light in the blue region of the spectrum, which can give the atmosphere a brown hue 66 and has thus been dubbed the Atmospheric Brown Cloud (9). The presence of these light 67 absorbing aerosols, and in particular those that can take on a dark hue against a light colored 68 background, suggest that the deposition of ambient particulate matter may be playing a role in 69 the discoloration of the outer white marble surfaces of the Taj Mahal.

70

71 Experimental Methods

72

Ambient Particulate Sampling and Analyses. In order to determine the influence of PM on the 73 74 Taj Mahal, ambient aerosol sampling was conducted for a roughly one-year period beginning on 75 Nov. 5, 2011 and continuing through June, 2012 just prior to the monsoon season. Filters were collected every 6th day each month for both PM_{2.5} (fine particulate matter having diameters less 76 77 than 2.5 µm) and total suspended particulate matter (TSP), and analyzed for major anions, 78 organic (OC) and elemental carbon (EC), and trace elements. The PM2.5 cut-point was 79 established using an upstream cyclone, while the TSP directly sampled ambient air. Portions of 80 each filter were combined to make monthly composites that were extracted and analyzed for 81 source specific trace organic compounds using GCMS. The trace organic concentrations were

used to estimate source contributions to particulate organic carbon using chemical mass balance
(CMB) modeling *(10)*. Additional information on the sampling and chemical analyses is
presented in the Supplementary Material section (sections S1 and S2).

85

86 Marble Deposition Target Sampling and Analyses. In addition to ambient samples, several 87 pre-cleaned marble deposition targets (with dimensions 2 cm x 2 cm x 0.5 cm) were placed 88 outdoors within roughly 300 m of the main Taj Mahal dome. Both the air sampling equipment 89 and targets were located in a section of the Taj Mahal that was accessible only to staff of the 90 Archaeological Survey of India (ASI), and had very little foot traffic. Pre-cleaned marble 91 cuboids were fastened to Taj Mahal structures with double-sided tape at a variety of locations, 92 and exposed from April to June 2012. Some of the marble samples were placed horizontally and 93 others vertically. Prior to, and after exposure, the marble samples were placed in sealed, pre-94 cleaned petri dishes and stored in a freezer to avoid degradation of deposited particles.

95 Scanning Electron Microscopy (SEM) (LEO 1530, Carl Zeiss Microscopy) and Energy Dispersive X-ray (EDX) Spectroscopy (Oxford Instruments X_{max} detector) were carried out on 96 97 two horizontally facing marble targets. Images were taken at many different magnifications to 98 capture the particles having sizes ranging from 100 nm to 100 µm. The particle sizes and shapes 99 were accessed through SEM images using image processing in a Matlab program. EDX analyses 100 were carried out on the same marble targets on ~ 1000 particles. The information gained from the 101 SEM/EDX analyses allowed for the estimation of the particle number and surface area 102 concentration, and chemical composition as a function of area of the marble target. This 103 information was then used, as described in the next sections, to estimate the change in color of the marble surface. More information on the SEM/EDX analyses is given in the SupplementaryMaterial Sections S1 and S2.

106

Linking Deposited Particles to Marble Surface Color. In order to estimate the impact of particles deposited to the marble substrate on the perceived color change of the surface, we developed an approach that estimates the influence of deposited particles on wavelengthdependent surface reflectance. The method builds on previous work that estimated the influence of particles deposited to plant leafs on available photosynthetically active radiation *(11)*. First, we use SEM/EDX analyses of particles deposited to the marble targets to estimate the optical depth of deposited particles as a function of wavelength (τ_{λ}) as follows:

114
$$\tau_{\lambda} = \frac{\pi}{4} \sum_{i=1}^{n} Ac_i D_{p,i}^2 \left[Q_{s\lambda} + Q_{a\lambda} \right]_i$$
(1)

115 Where Ac is the areal particle number concentration (number of particles deposited per area of 116 the marble surface) for each size bin, i, D_p is the particle diameter for deposited particles, and 117 $Q_{s\lambda}$ and $Q_{a\lambda}$ are the wavelength dependent Mie scattering and absorption efficiencies that are 118 determined based on particle size and composition.

119

120 After estimating the optical depth, the wavelength dependent single scattering albedo, ω_{λ} (ratio 121 of light scattering to extinction) is estimated as:

122
$$\omega_{\lambda} = \frac{\sum_{i=1}^{n} Ac_{i} D_{p,i}^{2} Q_{s\lambda,i}}{\sum_{i=1}^{n} Ac_{i} D_{p,i}^{2} [Q_{s\lambda} + Q_{a\lambda}]_{i}}$$
(2)

ACS Paragon Plus Environment

The single scattering albedo is a key parameter that determines the relative amount of light absorption that occurs over the white marble surface. For white, scattering only particles the single scattering albedo is near 1.0 and the surface reflectance of a white surface will not change.

127 The change in the surface reflectance of the white marble surface is calculated using SBDART, a 128 radiative transfer model *(12)* with input values including τ_{λ} and ω_{λ} estimated from equations 1, 129 and 2 as well as the asymmetry parameter (relative amount of light scattering in the forward 130 direction) as a function of wavelength using Mie theory *(13)*.

131

132 In order to determine the perceived color change of the white marble surface based on particle 133 deposition, and the related change in spectral surface reflectance we used the model described by 134 D'Andrade and Romney (14) to convert spectral reflectance to perceived color in the Munsell 135 color system. The Munsell color system is based on three components that include the value 136 (lightness/darkness), hue (color), and chroma (purity/saturation). The model used to estimate the 137 perceived color of the marble surface with deposited particles uses the spectral reflectance from 138 the radiative transfer model of the marble surface loaded with particles to estimate the Munsell 139 color. The Munsell color estimate also takes into consideration the human eye response as a 140 function of wavelength of incident light (the approach is described in more detail in the 141 supplementary information, section S3).

142

143 **Results and Discussion**

Ambient Particulate Concentrations. Figure 2a shows the average mass concentrations of particulate organic carbon mass (OM), ions, dust and elemental carbon (EC) for both TSP and

146 PM_{25} over the sampling 147 period. The mean daily 148 concentrations (and standard 149 deviations) of both TSP and 150 PM_{2.5} are 135 (55) and 60 (39) μ gm⁻³. The values are 151 152 significantly higher than the 153 annual World Health 154 Organization (WHO) PM 155 guidelines for PM_{10} and $PM_{2.5}$ of 20 μ gm⁻³ and 10 156 μgm^{-3} highlighting the poor 157 158 air quality in the region. The 159 fraction of particulate matter 160 greater than 2.5 μ m is ~60%, 161 and is due in large part to the 162 coarse mode dust that 163 increases from 15% of the 164 PM_{2.5} mass fraction to 30%

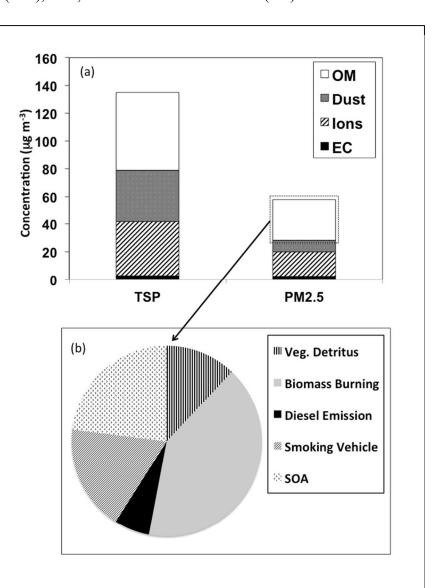
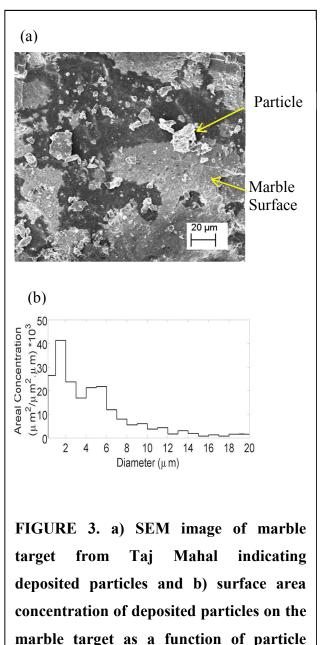


FIGURE 2. The general chemical composition of (a) Total Suspended Particulate matter (TSP) and fine particulate matter ($PM_{2.5}$), and (b) source apportionment of $PM_{2.5}$ organic mass (OM) based on filter sampling at the Taj

165 of the TSP mass. In addition to dust, other 166 PM components that absorb light in the 167 visible spectrum, and hence have the 168 possibility to influence the color of the outer 169 white marble surfaces, are elemental carbon 170 (EC) that is responsible for 2% of the TSP 171 mass, as well as OM that accounts for 39% of 172 the TSP mass. Estimates of the sources of 173 OM in the PM_{2.5} mass fraction shown in 174 Figure 2b indicate that biomass burning, a 175 known source of BrC, is responsible for 176 roughly half of the OM with significant 177 contributions from vehicular emissions. It 178 should be pointed out that biomass burning 179 OM can be from a variety of activities 180 including the combustion of wood and dung, 181 crop residue, and the burning of trash and 182 refuse that is ubiquitous in the region. The 183 prevalence of light absorbing aerosols in



Agra (i.e. elemental and organic carbon, and dust) suggests that PM deposition to the white marble surfaces may be responsible for the observed discoloration of the outer Taj Mahal structures including the famous Taj Mahal dome.

188 Size and Composition of Particles Deposited to Marble Targets. Figure 3a shows a Scanning 189 Electron Microscopy (SEM) image of a marble deposition target that was placed horizontally at 190 the Taj Mahal, and was exposed for a roughly two month period during the pre-monsoon season 191 of 2012. The surface area concentration (surface area of particles per unit marble surface area per 192 micron) distribution of particles as a function of particle size deposited to the marble target 193 (Figure 3b) shows peaks in particle size at roughly 1-2 μ m, and an additional mode at 4-5 μ m. 194 The fraction of the surface covered by particles is estimated to be $\sim 30\%$. Approximately 70% of 195 the deposited particle surface area is for particles having diameters greater than 2 μ m, indicating 196 that a large fraction of the particle surface area concentration is due to the deposition of coarse 197 particles. The dominance of coarse particles is due to both the relatively high concentration of 198 dust particles measured in Agra, as well as the fact that the dry deposition velocity of coarse (~ 5 199 μ m) particles is roughly 100 times greater than that of accumulation mode (~1.0 μ m) aerosol 200 particles (15). It is likely that dry deposition is the dominant mode of particle transport to the Taj 201 surface given the relatively low amount of precipitation in Agra during the Fall through Spring, 202 when particle loadings are high and the summer monsoon rainfall is not occurring. It is also 203 important to note that water insoluble particles (such as dust, BC, and a fraction of OM) are 204 likely not easily removed from the Taj Mahal surface by precipitation wash-off once deposited. 205 This is based on similar observations of the build-up of water insoluble particles on leaf surfaces 206 observed in the Yangtze delta region of China, an area that also experiences high PM loadings 207 (11).

208

The EDX analyses indicate that more than 70% of particles are primarily crustal in origin, with spectra dominated by crustal elements. The crustal particles were typically in the coarse (having

diameters greater than ~ 2-3 μ m) size fraction. Particles having diameters less than 2 μ m, which account for roughly 30% of the deposited particle surface area, also contained significant amounts of carbon likely from the sources of OM highlighted in CMB results in Figure 2b. More detailed information on the EDX analyses, including example spectra of particle EDX analyses are given in the supplementary section (S2).

216

217 **Deposited Particles and Perceived Color**. To estimate τ_{λ} we assume that particles having 218 diameters less than 3 µm are composed primarily of light absorbing organic carbon (BrC) with 219 wavelength dependent refractive indices reported by Liu et al. (16). This assumption is based on 220 the fact that ambient filters indicated that roughly half of the PM_{2.5} was carbonaceous in nature, 221 combined with the fact that we did not observe the presence of major ion related elements (i.e. S) 222 deposted to the marble targets, but did see a dominance of carbon particles in the less than 3 µm 223 particle sizes. We also assume that 10% of the particles less than 3 μ m are black carbon (BC) 224 particles with a refractive index commonly used for soot particles (13). It is likely that we 225 overestimate the influence light absorption by BrC and BC since we assume that all of the 226 particles less than 3 µm are carbonaceous and not elemental or ionic in nature. Aerosol optical 227 depth contributions at 400 nm for dust, BrC and BC are estimated to be 0.222, 0.144 and 0.016 228 respectively, highlighting the importance of light extinction by all three components. The single scattering albedo at 400 nm and 700 nm is estimated to be 0.64 and 0.95, indicating that a 229 230 significant amount of light absorption occurs at near-ultraviolet wavelengths preferentially to 231 that at the longer 700 nm wavelength.

232

Figure 4a shows the surface reflectance of a pure marble surface as well as the estimated surface reflectance for several cases including the influence from each light absorbing particulate

235 component separately (BC, BrC and 236 and the case when dust). all 237 components are combined. As seen 238 in the plot, BC absorbs across all 239 wavelengths evenly, with brown 240 showing preferential carbon 241 absorption at shorter wavelengths 242 near 400 nm. Dust substantially 243 reduces the surface reflectance at all 244 wavelengths, and preferentially so at 245 shorter wavelengths due to the 246 presence of hematite, which absorbs 247 blue wavelengths. at When 248 combined, dust, BrC, and BC are 249 estimated to substantially alter the 250 surface reflectance with enhanced 251 absorption (i.e. lower values of 252 surface reflectance) at shorter 253 wavelengths.

254

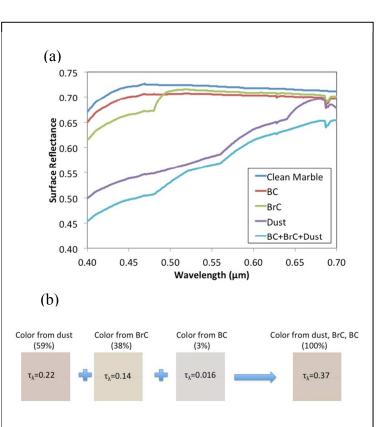


FIGURE 4. (a) Estimated marble target surface reflectance for a clean surface and surface area coverage of particles based on Fig. 3 for black carbon (BC), Brown Carbon (BrC), Dust and all particles (BC+BrC+Dust) (b) Change in color of white marble surface for dust, BrC, BC separately and combined. Values in parenthesis represent fraction of total surface area concentration contributed by each component with AOD values estimated by Eqn. 1 at 400 nm

255 Figure 4b shows the estimated change in color of the white surface for deposition of the light 256 absorbing particles measured in Figure 3b over the two-month period the targets were exposed. 257 Results indicate that separately each component contributes to the color change of the Taj Mahal 258 white marble surfaces. For BC alone (which we estimate to account for 3% of the total particle 259 surface area) the color change results in a greyish color given that the change in surface 260 reflectance is proportionally similar at each wavelength. Both BrC (~30% of the total particle 261 surface area) and dust influence color with preferential absorption in the UV resulting in 262 yellowish-brown hues. When combined, the perceived color of the surface shifts towards darker 263 shades of yellow-brown. It should be pointed out that our sample targets were mounted for a 264 relatively short time (~2 months) compared to the typical time between cleanings of the outer Taj 265 Mahal surfaces (several years), and therefore it may be expected that the perceived color of the 266 marble target would be attenuated compared to that of the white surfaces of the Taj Mahal. 267 Indeed the marble target surface did appear somewhat lighter in color as compared to the color 268 estimates in Figure 4b, although qualitatively were similar. There are several uncertainties in 269 estimating perceived color including the loading, particle size, and optical properties. Analyses 270 (included in section S3) suggest that results are moderately sensitive to both particle loading and 271 size. For example, assuming an uncertainty of 50% for aerosol loading only moderately 272 influences the perceived color and does not change the conclusions that both dust and 273 carbonaceous particles contribute to the perceived color change of the Taj Mahal. Overall, the 274 results indicate that light absorbing particles play an important role in the discoloration of the Taj 275 Mahal surface and that dust, as well as BC and BrC that are primarily from biomass combustion, 276 trash/refuse burning, and mobile sources, all make significant contributions to the discoloration.

277 This work further suggests that the deposition of light absorbing particulate matter to both 278 natural and human-made surfaces results in a substantial discoloration in regions of high aerosol 279 loading. The discoloration impacts not only cultural artifacts but also the aesthetics of the 280 environment through the modification of surface albedo, and hence perceived color. The 281 measurement/modeling approach developed in this paper allows surface color changes to be 282 estimated based on the relative amounts of light absorbing particles deposited to surfaces, and 283 can be used to develop future control strategies to prevent the discoloration of the environment 284 by particle deposition, that will also improve air quality.

Acknowlegements: This work was supported by grants from the Indo US Science and Technology Forum (IUSSTF) as well as from the US EPA (grant # RD83479901), and the NSF PIRE Grant 1243535. In addition, chemical analyses at IIT Kanpur were conducted by N. Rastogi and C.M. Shukla. We are also grateful for the efforts by the ASI Staff at the Taj Mahal.

Supporting information is associated with this paper and includes details related to filter and marble surrogate surface sampling (S1), chemical analyses on both the filters and marble surrogate surfaces (S2), as well as the approach to estimate perceived color based on surface reflectance (S3). The information is available free of charge via the internet at http://pubs.acs.org.

294

295 **References**

296

Sharma, P.K; Gupta, H.O., Dust pollution at the Taj Mahal-A case study, *Proceedings of RILEM/UNESCO Congress*, Ed. M.J. Thiel **1993**, pp. 11-18

2. B. B. Hicks, K. Manju, Marble discoloration at the Taj Mahal: A proposed explanation **1987**, in *ICOMOS 8th General Assembly and International Symposium*. (Washington, D.C,
U.S.A.), pp. 325-332.

302 3. Prasad, A. K.; Singh, R. P.; Singh, A., Seasonal climatology of aerosol optical depth over
303 the Indian subcontinent: trend and departures in recent years. *International Journal of Remote*304 *Sensing* 2006, *27*, (12), 2323-2329.

4. Dey, S.; Di Girolamo, L.; van Donkelaar, A.; Tripathi, S. N.; Gupta, T.; Mohan, M.,
Variability of outdoor fine particulate (PM2.5) concentration in the Indian Subcontinent: A
remote sensing approach. *Remote Sensing of Environment* 2012, *127*, 153-161.

<u>308</u> 5. Massey, D. D.; Kulshrestha, A.; Taneja, A., Particulate matter concentrations and their
 <u>309</u> related metal toxicity in rural residential environment of semi-arid region of India. *Atmospheric* <u>310</u> *Environment* 2013, 67, 278-286.

6. Kulshrestha, A.; Satsangi, P. G.; Masih, J.; Taneja, A., Metal concentration of PM2.5 and
PM10 particles and seasonal variations in urban and rural environment of Agra, India. *Science of the Total Environment* 2009, *407*, (24), 6196-6204.

314 7. Satsangi, P. G.; Kulshrestha, A.; Taneja, A; Rao, P.S.P., Measurements of PM10 and PM2.5
315 aerosols in Agra, a semi-arid region of India, *Indian Journal of Radio & Space Physics* 40, 2011,
316 <u>203-210.</u>

8. Arola, A.; Schuster, G.; Myhre, G.; Kazadzis, S.; Dey, S.; Tripathi, S. N., Inferring
absorbing organic carbon content from AERONET data. *Atmospheric Chemistry and Physics*2011, 11, (1), 215-225.

320	9. Ramanathan, V.; Chung, C.; Kim, D.; Bettge, T.; Buja, L.; Kiehl, J. T.; Washington, W. M.;
321	Fu, Q.; Sikka, D. R.; Wild, M., Atmospheric brown clouds: Impacts on South Asian climate and
322	hydrological cycle. Proceedings of the National Academy of Sciences of the United States of
323	America 2005 , <i>102</i> , (15), 5326-5333.
324	10. Schauer, J. J.; Rogge, W. F.; Hildemann, L. M.; Mazurek, M. A.; Cass, G. R.; Simoneit, B.
325	R. T., Source apportionment of airborne particulate matter using organic compounds as tracers.
326	Atmospheric Environment 1996, 30, (22), 3837-3855.
327	11. Bergin, M. H.; Greenwald, R.; Xu, J.; Berta, Y.; Chameides, W. L., Influence of aerosol
328	dry deposition on photosynthetically active radiation available to plants: A case study in the
329	Yangtze delta region of China. Geophysical Research Letters 2001, 28, (18), 3605-3608.
330	12. Ricchiazzi, P.; Yang, S. R.; Gautier, C.; Sowle, D., SBDART: A research and teaching
331	software tool for plane-parallell radiative transfer in the Earth's atmosphere. Bulletin of the
332	American Meteorological Society 1998, 79, (10), 2101-2114.
<u>333</u>	13. Carrico, C. M.; Bergin, M. H.; Xu, J.; Baumann, K.; Maring, H., Urban aerosol radiative
334	properties: Measurements during the 1999 Atlanta Supersite Experiment. Journal of Geophysical
335	Research-Atmospheres 2003, 108, (D7), 17.
336	14. D'Andrade, R. G.; Romney, A. K., A quantitative model for transforming reflectance
337	spectra into the Munsell color space using cone sensitivity functions and opponent process

weights. Proceedings of the National Academy of Sciences of the United States of America 2003,
100, (10), 6281-6286.

- 340 15. Slinn, W. G. N., Predictions for particle deposition to vegatiative Canopies. *Atmospheric*
- 341 Environment 1982, 16, (7), 1785-1794.
- 342 16. Liu, J.; Bergin, M.; Guo, H.; King, L.; Kotra, N.; Edgerton, E.; Weber, R. J., Size-resolved
- 343 measurements of brown carbon in water and methanol extracts and estimates of their
- 344 contribution to ambient fine-particle light absorption. *Atmospheric Chemistry and Physics* 2013,
- *3*45 *13*, (24), 12389-12404.

346

347

348